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## Improvements to Disordered Rocksalt Li-Excess Cathode Materials

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### **Presentation Outline**



- The LIB Cathode Market
  - $\circ$  The Herd
  - $\circ$  The Opportunity
- Disordered Rocksalt
  - How it's Different
  - Performance Challenges
  - Development Progress
- Summary

### EV and E-Bus Markets are Growing Rapidly





#### The transportation market is driving enormous LIB demand

### Lithium-Ion Cathode Demand – 2010 vs. 2017





### NMC and LFP use have grown dramatically since 2010

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### 2017 Cathode Demand (Tonnes) by Market





#### NMC is the heavy favorite for automotive applications

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### Lithium-Ion Cathode Demand – 2010 vs. 2025





#### NMC is forecast to take over 70% of the global market by 2025

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### **Cobalt Concerns**





#### What can be done to reduce our reliance on cobalt?

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### The Move to Higher Nickel NMC's



#### More nickel, less cobalt

- Strong move toward higher nickel content NMC's
- Reduction of Co content from 33% to less than 10%
- And even higher nickel versions are being considered (>90%)



#### NMC 811 is one of the world's most popular near-term R&D targets

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### Energy Improvements are Possible, but Modest



#### Increase energy

- Cell energies have increased 3% per year, but are near cathode theoretical limits
- Increased voltage leads to higher energies...
- ...but introduces electrolyte stability and gas generation challenges



Source: Avicenne Energy

Are there alternatives to NMC...with higher energy and no Cobalt?

### **Cathode Material Energy**





A gap exists between today's LIB energies and post-LIB technologies

### **Cathode Material Energy**





Disordered rocksalt is a Co-free alternative to NMC with ~40-60% more energy

### Cathode Material Cost per kWh





#### DR offers a tremendous \$/kWh value proposition

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### Ordered vs. Disordered Structures





Today's cathodes are mostly ordered structures:

- Li sites and pathways separated from TM sublattice
- Provides stability (good cycle life)
- TM layers provide electron storage capacity

In a disordered rock salt:

- Both Li and TM occupy a cubic close-packed lattice of octahedral sites
- Li diffusion occurs by hopping from one octahedral site to another via an intermediate tetrahedral site (*o-t-o* diffusion)
- Very small changes in lattice parameters during cycling

Disordered structures appear to have performance advantages

### Are Fast Kinetics Possible?





Multiple possible pathways for Li transport

0-TM pathway is energetically favored

Li content needs to be high for 0-TM percolation

#### Percolation network provides means of fast lithium transport

### **Rich Compositional Space**



Year	Author et al.	Material	Journal	capacity	
2014	J. Lee	Li <sub>1.211</sub> Mo <sub>0.467</sub> Cr <sub>0.3</sub> O <sub>2</sub>	Science	280 mAh/g	
2015	Chen	Li <sub>2</sub> VO <sub>2</sub> F	Adv. Energy Mater.	320 mAh/g	
2015	Yabuuchi	Li <sub>1.3</sub> Nb <sub>x</sub> M <sub>0.7-x</sub> O <sub>2</sub> (M = Mn, Ni, Co, Fe)	PNAS	300 mAh/g	
2015	Wang	Li <sub>1.25</sub> Nb <sub>0.25</sub> Mn <sub>0.5</sub> O <sub>2</sub>	Electrochem. Commun.	290 mAh/g	
2015	J. Lee	$\rm Li_{1.2}Ni_{0.333}Ti_{0.333}Mo_{0.133}O_{2}$	Energy Environ. Sci.	230 mAh/g	
2015	Glazier	Li <sub>1+x</sub> Ti <sub>2x</sub> Fe <sub>1-3x</sub> O <sub>2</sub>	Chem. Mater.	250 mAh/g	N
2016	Freire	Li <sub>4</sub> Mn <sub>2</sub> O <sub>5</sub>	Nat. Mater.	350 mAh/g	1
2016	Yabuuchi	Li <sub>1.3</sub> Nb <sub>0.3</sub> V <sub>0.4</sub> O <sub>2</sub>	Chem. Commun.	270 mAh/g	]
2017	Yabuuchi	Li <sub>4/3</sub> Mo <sup>6+</sup> 2/9Mo <sup>3+</sup> 4/9O <sub>2</sub>	ACS Energy Letters	330 mAh/g	
2017	J. Lee	$Li_{1.15}Ni_{0.45}Ti_{0.3}Mo_{0.1}O_{1.85}F_{0.15}$	Nat. Commun.	250 mAh/g	
2017	Yabuuchi	$LiMoO_{2-x}$ - $LiF$ ( $0 \le x \le 2$ )	Journal of Power Sources	320 mAh/g	
2018	Bruce	Li <sub>2</sub> MnO <sub>2</sub> F	Energy Environ. Sci.	280 mAh/g	
2018	J. Lee	Li <sub>2</sub> Mn <sub>2/3</sub> Nb <sub>1/3</sub> O <sub>2</sub> F / Li <sub>2</sub> Mn <sub>1/2</sub> Ti <sub>1/2</sub> O <sub>2</sub> F	Nature	320 mAh/g	
2018	Kitchaev	Li-Mn(II)-V(IV)-O-F	Energy Environ. Sci.	310 mAh/g	Ce

NCM622 = 185 mAh/g

Ceder, *IMLB* (2018)

#### Disordered rocksalts offer a rich compositional space to explore

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### **Nb-based Materials**





Many cation ordered/disordered rocksalt phases on tie-line between MeO and LiO, including Li<sub>3</sub>Nb<sup>5+</sup>O<sub>4</sub> However,  $Li_3Nb^{5+}O_4$  has poor capacity due to poor electronic conductivity, with no electrons in a conduction band (4d<sup>0</sup> configuration for Nb<sup>5+</sup>)

Nb phases provide a particularly promising area for exploration

### **Disordered Material Capacity**







TM substitution for some of the Nb<sup>5+</sup> and Li<sup>+</sup> induces electronic conductivity in  $Li_3NbO_4$  – yielding capacities beyond limit of redox for TM's Electrochemical performance of  $Li_{1.3}Nb_{0.3}Mn_{0.4}O_2$  (1.5 – 4.8V, 10 mA/g, 60°C) [0.43Li<sub>3</sub>NbO<sub>4</sub>-0.57LiMnO<sub>2</sub>]

Observed capacity is higher than theoretical based on TM redox

### **Disordered Rocksalt Challenges**





- Most literature results are at C/40 and at elevated temperature
- 2. Low discharge voltage
  - Capacity above 3.0V is less than 250 mAh/g
- 3. Poor cycle life
  - Mainly reported in half cells



Cycle number

#### Disordered materials have high energy, but other metrics need improvement

Τ

Phase

 $\sim$ 

Phase

### **High Throughput Material Development**





#### Success depends upon finding the right *combination* of materials and processes

## Conductivity Improvement via Composition



- Conductivity sensitive to both Li content and cation mixing
- For given TM's, Li content can be studied to optimize conductivity



Cedar, *Science*, 343, 2014

#### Li<sup>+</sup> percolation can be optimized by changing composition

### Conductivity Improvement via Composition





Li content can be optimized, but other improvements are needed

### Effect of **Dopants** on Performance





- $\circ$  d-band of TM overlapping p-band of oxygen
- Prevents O<sub>2</sub> release from material

100

DR//Li 4.8-1.5V

0.10

C/10 Capacity

M1M3

MIMS

300 250



Complex DOE's required to optimize both lithium and dopant compositions

Multiple

dopants

### **Carbon Coating Can Improve Rate Performance**





**1C Capacity** 

Performance is very sensitive to both base composition and carbon coating

### **Development Progress**





#### Complex DOE's enable systematic improvements

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### **Development Progress**



- Wildcat improved performance via:
  - o Elemental ratios
  - Doping
  - $\circ$  Carbon coating
- Specific energy of best materials at 30°C outperform literaturereported materials at 55°C
- Phase 2 development will focus on improved cycle life and voltage plateau



Wildcat's disordered rocksalt has energy density of ~975 Wh/kg at room temperature

### Summary



- Auto LIB market is driving tremendous growth in NMC demand
- Cobalt's pricing and labor concerns and the desire for more energy are driving a strong move to higher nickel NMC's
- Disordered rocksalt offers a compelling alternative to NMC, with higher energy, lower \$/kWh and no reliance on cobalt
- Disordered materials inherently offer more energy than layered materials, yet performance challenges exist: rate, cycle life, voltage plateau
- A vast DR compositional region remains unexplored with many opportunities to further improve performance
- High throughput is an excellent means of rapidly evaluating hundreds of new compositions
- DR is young, there are large improvements ahead!

### Wildcat's Business Model





Wildcat Website: www.wildcatdiscovery.com

High Throughput Video: https://youtu.be/Eu1nvrCzJWQ

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# Thank you for your attention!

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